SEQUENTIAL MICHAEL-MICHAEL-MICHAEL-RING CLOSURE REACTIONS FOR

HIGH-YIELD, ONE-POT, 4-COMPONENT COUPLING

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Summary: An extraordinarily easy method is presented for connecting four components leading efficiently to functionalized cyclohexenes, cyclohexanols, cyclodecenones, and aromatic systems.

We have developed one-pot, sequential, <u>Michael-Michael-Ring</u> Closure (MIMIRC) reactions for effective construction of various polyfunctionalized cyclohexyl units by coupling of <u>three</u> 2-carbon components.¹⁻³ Controlled coupling of <u>four</u> components in one reaction vessel is clearly an even more demanding and potentially more useful achievement.⁴ We now report successful attainment of this goal as illustrated generally in eq. 1 involving formation of the four bonds a-d in structure <u>1</u> and leading to a variety of richly functionalized carbocycles in excellent yields.



Tris(ethylthio)methyllithium⁵ added in a Michael fashion to 2-cyclohexenone to generate an intermediate enolate ion which reacted with two equivalents of vinyltriphenylphosphonium bromide (VTB) under dilute (motor-driven syringe) conditions to give cyclohexannulation product 2 after treatment with potassium hydroxide (to convert the initially-formed phosphonium salt into a more easily isolated phosphine oxide).^{1,6} This 4-component coupling reaction (eq. 2) proceeded with an average yield of 83% per each of the four new carbon-carbon bonds!

Trialkyltinlithium reagents⁷ added in a Michael fashion to 2-cyclohexenone, and the resultant enolate intermediate underwent double Michael addition to two equivalents of methyl acrylate⁸ or methyl α -bromoacrylate leading to ester enolate intermediates which spontaneously







underwent intramolecular aldol cyclizations to form polysubstituted cyclohexanols 3a and 3b, each as a mixture of only two diastereomers^{2,6} (Scheme I). A motor-driven syringe was not This 4-component cyclohexannulation reaction proceeded with a stunning average required. yield of 93-94% per each new C-C bond! The very rich functionality of annulation products 3a and <u>3b</u> allows many different further chemical manipulations. For example, the major diastereomer of γ -hydroxystannane 3a was oxidatively cleaved by lead tetraacetate 9 to form highly functionalized trans-cyclodecenone 4 in 60% yield;⁶ the minor diastereomer of γ hydroxystannane 3a was likewise oxidatively cleaved to produce the corresponding ciscyclodecenone in 56% yield.⁶ This efficient, 2-step conversion of cyclohexenone into a stereodefined, polyfunctionalized cyclodecene system may find use in rapid synthetic entry into the germacrane class of sesquiterpenes.¹⁰ Oxidation of dibromo- γ -hydroxystannane 3b with chromium trioxide⁷ afforded dibromo- β -hydroxyketone 5 (50% yield) which was dehydrated and doubly dehydrobrominated (thionyl chloride/pyridine, $0^{\circ}C \rightarrow reflux$, l hour) to form aromatic ketone 6 (64% yield);⁶ this overall sequence represents a short, easy and direct means of transforming cyclohexenone into a regiospecifically tetrasubstituted aromatic system having a 1,3,5-tricarbonyl substitution pattern.¹¹ In this process, the cycloxenone carbonyl group has effectively undergone a 1,3-carbonyl transposition, 1^2 and the original cyclohexenone -COCHunit has undergone overall annulation of a m-dicarboxylated aromatic ring.

All of the 4-component cyclohexannulations in eq. 2 and in Scheme I involve three different components. An even greater challenge is to couple four different components in a controlled manner. 2-Cyclohexenone reacted sequentially with tri-n-butyltinlithium, ethyl vinyl ketone, and methyl acrylate to produce cyclohexannulation product $\frac{7}{2}$ (eq. 3)⁶ with an average yield of 86% per new bond! This result is a truly dramatic demonstration of the great synthetical potential of this multi-component coupling process.

Although the sequential <u>Michael-Michael-Michael-Ring</u> <u>Closure</u> (MIMI-MIRC)^{1,13} reactions shown here are illustrated only with 2-cyclohexenone as the initial Michael acceptor, our experience already indicates that other α -enones will be usable as well. We are actively pursuing further applications.

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- 4. For recent examples, see Deshpande, M. N., Jawdosiuk, M., Kubiak, G., Venkatachalam, M., Weiss, U. and Cook, J. M., J. Am. Chem. Soc., 1985, 107, 4786; and Bertz, S. H., J. Org. Chem., 1985, 50, 3585.
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